

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

DECLARATION OF ACCURACY OF TRANSLATION

The undersigned translator, Sachiko Miki of TSUKUNI & ASSOCIATES having an office at 1-22-12, Toranomom, Minato-ku, Tokyo, Japan states that:

- (1) I am fully conversant both with the Japanese and English languages.
- (2) I have carefully compared the attached English-language translation of Japanese Patent Application Number 2002-287071, filed September 30, 2002 with the original Japanese-language patent application.
- (3) The translation is, to the best of my knowledge, and belief, an accurate translation from the original into the English language.

Date: April 8, 2008

S. Miki
Sachiko Miki

[Document name] Specification

[Title of the invention] RESIN COMPOSITION FOR PRINTED WIRING BOARD AND VARNISH, PREPREG AND METAL CLAD LAMINATED BOARD USING THE SAME

[Scope of claim for patent]

[Claim 1] A resin composition for printed wiring board characterized in being obtained by using (A) a cyanate ester compound having 2 or more cyanate groups in the molecule and/or a prepolymer thereof, (B) a monovalent phenol compound, (C) a polyphenylene ether resin, and (D) an epoxy resin containing at least one kind of an epoxy resin having a biphenyl structure in the molecule.

[Claim 2] The resin composition for printed wiring board according to Claim 1, which comprises Components (A) to (D).

[Claim 3] The resin composition for printed wiring board according to Claim 2, which comprises 2 to 60 parts by weight of Component (B), and 10 to 250 parts by weight of Component (D) based on 100 parts by weight of Component (A).

[Claim 4] The resin composition for printed wiring board according to Claim 1, which comprises a phenol-modified cyanate ester oligomer obtainable by reacting Component (A) with Component (B), and Component (C) and Component (D).

[Claim 5] The resin composition for printed wiring board according to Claim 4, wherein the phenol-modified cyanate ester oligomer is a phenol-modified cyanate ester oligomer obtainable by reacting Component (A) with Component (B) in the presence of Component (C).

[Claim 6] The resin composition for printed wiring board according to Claims 4 or 5, wherein the phenol-modified cyanate ester oligomer is a phenol-modified cyanate ester oligomer obtainable by reacting 100 parts by weight of Component (A) and 2 to 60 parts by weight of Component (B).

[Claim 7] The resin composition for printed wiring board according to Claims 4 or 5, wherein the phenol-modified cyanate ester oligomer is a phenol-modified cyanate ester oligomer obtainable by reacting 100 parts by weight of Component (A), 0.4 parts by weight or more and less than 60 parts by weight of Component (B), and Component (B) is additionally contained in a total amount of 2 to 60 parts by weight which is the sum of the amount with Component (B) to be used for formation of the phenol-modified cyanate ester oligomer.

[Claim 8] The resin composition for printed wiring board according to any one of Claims 4 to 7, wherein Component (C) is 5 to 300 parts by weight and Component (D) is 10 to 250 parts by weight based on 100 parts by weight Component (A).

[Claim 9] The resin composition for printed wiring board according to Claim 1, which comprises an epoxy/phenol-modified cyanate ester oligomer obtainable by reacting Component (A) with Component (B) and Component (D), and Component (C).

[Claim 10] The resin composition for printed wiring board according to Claim 9, wherein the epoxy/phenol-modified cyanate ester oligomer is an epoxy/phenol-modified cyanate ester oligomer obtainable by reacting Component (A), Component (B) and Component (D) in the presence of Component (C).

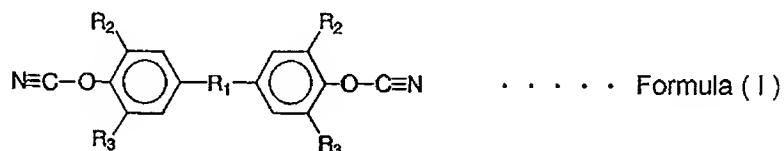
[Claim 11] The resin composition for printed wiring board according to Claims 9 or 10, wherein the epoxy/phenol-modified cyanate ester oligomer is a epoxy/phenol-modified cyanate ester oligomer obtainable by reacting 100 parts by weight of Component (A), 0.4 parts by weight or more and less than 60 parts by weight of Component (B), and Component (D) is contained in an amount of 10 to 250 parts by weight, and Component (B) is additionally contained in a total amount of 2 to 60 parts by weight which is the sum of the amount with Component (B) to be used for formation of the

epoxy/phenol-modified cyanate ester oligomer.

[Claim 12] The resin composition for printed wiring board according to any one of Claims 9 to 11, wherein Component (C) is 5 to 300 parts by weight based on 100 parts by weight of Component (A).

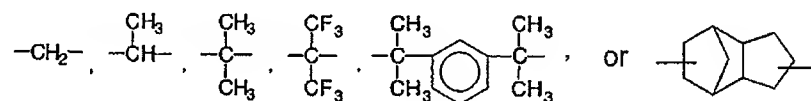
[Claim 13] The resin composition for printed wiring board according to any one of Claims 1 to 12, wherein Component (A) is at least one selected from the group consisting of a cyanate ester compound represented by the formula (I):

[Formula 1]



(wherein R₁ represents

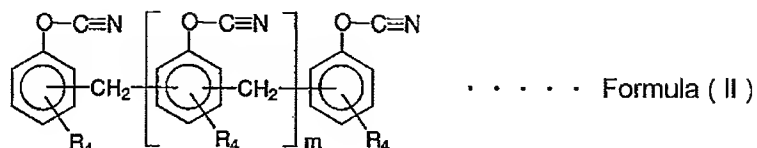
[Formula 2]



R₂ and R₃ each represent a hydrogen atom or an alkyl group having 1 to 4 carbon atoms, and each may be the same or different from each other)

and a cyanate ester compound represented by the formula (II):

[Formula 3]



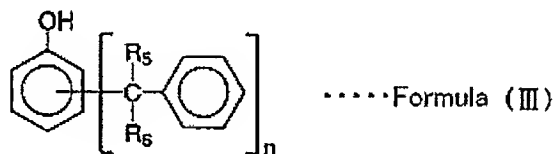
(wherein R₄ represents a hydrogen atom or an alkyl group having 1 to 4 carbon atoms, m represents an integer of 1 to 7.)

and a prepolymer thereof.

[Claim 14] The resin composition for printed wiring board according to Claim 13, wherein Component (A) is at least one selected from the group consisting of 2,2-bis(4-cyanatophenyl)propane, bis(4-cyanatophenyl)ethane, bis(3,5-dimethyl-4-cyanatophenyl)methane, 2,2-bis(4-cyanatophenyl)-1,1,1,3,3,3-hexafluoropropane, α,α' -bis(4-cyanatophenyl)-m-diisopropylbenzene, a cyanate ester compound of a phenol-added dicyclopentadiene polymer, a phenol novolac type cyanate ester compound and a cresol novolac type cyanate ester compound and a prepolymer thereof.

[Claim 15] The resin composition for printed wiring board according to any one of Claims 1 to 14, wherein Component (B) is at least one selected from the group consisting of phenol compounds represented by the formula (III):

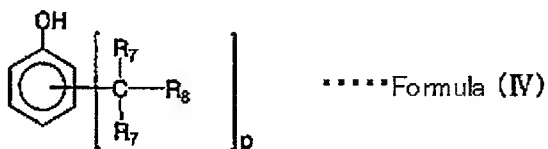
[Formula 4]



(wherein R_5 and R_6 each represent a hydrogen atom or a methyl group, each may be the same or different from each other, and n represents an integer of 1 to 3)

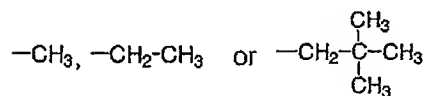
and a monovalent phenol compound represented by the formula (IV):

[Formula 5]



(wherein R₇ represents a hydrogen atom or a methyl group, R₈ represents

[Formula 6]



and p represents an integer of 1 or 2.)

[Claim 16] The resin composition for printed wiring board according to Claim 18, wherein Component (B) is at least one selected from the group consisting of p-(α -cumyl)phenol, mono-, di- or tri-(α -methylbenzyl)phenol, p-tert-butylphenol, 2,4- or 2,6-di-tert-butylphenol, p-tert-amylphenol and p-tert-octylphenol.

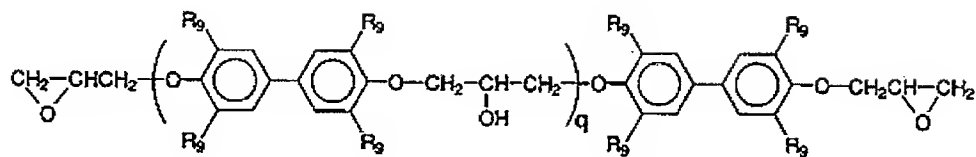
[Claim 17] The resin composition for printed wiring board according to any one of Claims 1 to 16, wherein Component (C) is an alloy polymer comprising poly(2,6-dimethyl-1,4-phenylene)ether and an alloy polymer comprising polystyrene, poly(2,6-dimethyl-1,4-phenylene)ether and a styrene-butadiene copolymer, which polymer contains containing 50% or more of the poly(2,6-dimethyl-1,4-phenylene)ether.

[Claim 18] The resin composition for printed wiring board according to any one of Claims 1 to 17, wherein Component (D) of the present invention is a component in which at least one of the epoxy resin containing a biphenyl structure and at least one of an epoxy resin other than the epoxy resin containing a biphenyl structure are combined, the other epoxy resin is a phenol novolac type epoxy resin, a cresol novolac type epoxy resin, a brominated bisphenol A type epoxy resin and a brominated phenol novolac type epoxy resin.

[Claim 19] The resin composition for printed wiring board according to any one of Claims 1 to 18, wherein the epoxy resin having a biphenyl structure in the molecule is at least one selected from the group consisting of an epoxy

resin represented by the formula (V):

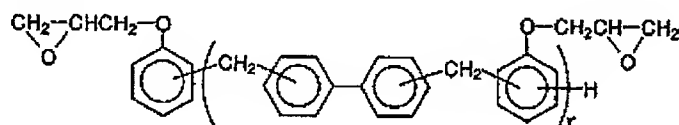
[Formula 7]



.....Formula (V)

(wherein R₉ each represent a hydrogen atom or a methyl group, q represents an integer of 0 to 6) and an epoxy resin represented by the formula (VI):

[Formula 8]



.....Formula (VI)

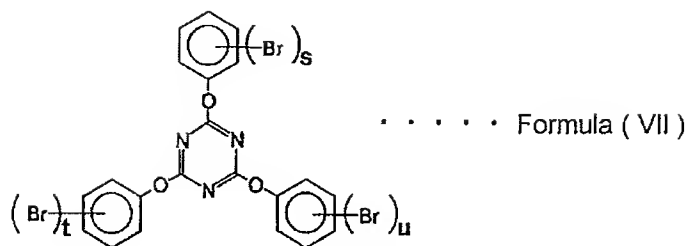
(wherein r represents an integer of 1 to 5.)

[Claim 20] The resin composition for printed wiring board according to any one of Claims 1 to 19, which further comprises (E) a flame retardant.

[Claim 21] The resin composition for printed wiring board according to Claim 20, wherein Component (E) is 5 to 100 parts by weight based on the total amount of Components (A) to (D) as 100 parts by weight.

[Claim 22] The resin composition for printed wiring board according to Claims 20 or 21, wherein Component (E) is at least one kind selected from the group consisting of 1,2-dibromo-4-(1,2-dibromoethyl)cyclohexane, tetrabromocyclooctane, hexabromocyclododecane, bis(tri-bromophenoxy)ethane, and a brominated polyphenylene ether, brominated polystyrene, and a brominated triphenylcyanurate flame retardant represented by the formula (VII):

[Formula 9]



(wherein s , t and u each represent an integer of 1 to 5, and each may be the same value or different from each other).

[Claim 23] The resin composition for printed wiring board according to any one of Claims 1 to 22, which further comprises an antioxidant.

[Claim 24] The resin composition for printed wiring board according to Claim 23, wherein Component (F) is 0.1 to 20 parts by weight based on the total amount of Components (A) to (D) as 100 parts by weight.

[Claim 25] The resin composition for printed wiring board according to Claims 23 or 24, wherein Component (F) is at least one kind selected from the group consisting of a phenol type antioxidant and a sulfur type antioxidant.

[Claim 26] A resin varnish for a printed wiring board obtainable by dissolving or dispersing the resin composition for printed wiring board according to any one of Claims 1 to 25 in a solvent.

[Claim 27] A prepreg for a printed wiring board which is obtainable by impregnating the resin composition for printed wiring board according to any one of Claims 1 to 25 or the resin varnish for a printed wiring board according to Claim 26 into a substrate, and drying at 80 to 200°C.

[Claim 28] A metal clad laminated board which is obtainable by laminating one or more of the prepreg for a printed wiring board according to Claim 27, laminating a metal foil on at least one surface thereof and pressurizing under heating.

[Detailed description of the invention]

[0001]

[Technical field to which the invention belongs]

The present invention relates to a resin composition for printed wiring board, and a varnish, a prepreg and a metal clad laminated board using the same. More specifically, it relates to a resin composition for printed wiring board which is used for electronic devices in which operating frequency exceeds 1 GHz, and a varnish, a prepreg and a metal clad laminated board using the same.

[0002]

[Prior art]

In recent years, in mobile communication devices represented by cellular phone or in net-work related electronic devices such as server, router, etc., it is required to transmit and treat huge amount of information with low loss and high speed, so that electric signals treated on the printed wiring board also become high frequencies. However, when a frequency is higher, electric signals likely decreased, so that it is required to use a material having low transmission loss for a printed wiring board to be used in these fields of art. That is, in a high frequency band region of 1 GHz or higher, a material having excellent dielectric characteristics represented by low values in dielectric constant and dielectric dissipation factor is required to be used.

[0003]

Also, in such background, as a resin composition for printed wiring board, a resin composition in which a cyanate ester excellent in dielectric characteristics and a polyphenylene ether are mixed and kneaded has been proposed (for example, see Patent Reference 1). However, in these resin compositions, when a formulation amount of the cyanate ester is larger, a dielectric dissipation factor tends to be high relative to the value of the dielectric constant. On the other hand, when a formulation amount of

the polyphenylene ether is increased to lower the dielectric dissipation factor, a melting viscosity of the resin composition becomes high whereby flowability thereof becomes insufficient, so that there are problems that molding property is worsened.

[0004]

Also, a resin composition in which an epoxy resin such as a bisphenol A type epoxy resin, a brominated bisphenol A type epoxy resin, a phenol novolac type epoxy resin and a cresol novolac type epoxy resin is formulated to a cyanate ester resin and a polyphenylene ether has been proposed (for example, see Patent Reference 2). However, in these resin compositions, there are problems that dielectric characteristics at high frequency band region do not yet satisfy with a required level due to the effects of components other than the cyanate ester resin and the polyphenylene ether. In particular, in a multi-layered printed wiring board to be used for wireless base station equipment use or high speed server, router, etc., temperature of the printed wiring board during operation of the device sometimes becomes a high temperature of 85 to 90°C, whereby there are possibility of causing significant problems that a dielectric constant is fluctuated due to change in temperature, or mismatching in impedance or transmission error accompanied by increase in transmission loss occurs when a dielectric dissipation factor becomes high. Accordingly, necessity for a resin composition for printed wiring board excellent in dielectric characteristics including temperature-dependency has been increased.

[0005]

Also, a resin composition excellent in heat resistance, molding property and workability as well as having good dielectric characteristics at high frequency band region has been proposed by the present inventors by modifying a cyanate ester resin with a specific monovalent

phenol compound to prepare a phenol-modified cyanate ester resin composition, whereby dielectric characteristics possessed by a cyanate ester resin have been more improved, and further a polyphenylene ether resin is formulated (for example, see Patent Reference 3 and 4). However, in these resin compositions, it has been required to further improve humidity resistance under sever conditions such as a pressure cooker test for a long period of time, etc. while maintaining good dielectric characteristics including temperature-dependency.

[0006]

[Patent Reference 1]

Patent Publication No. S 61-18937

[Patent Reference 2]

Patent Publication No. H 4-57696

[Patent Reference 3]

Unexamined Patent Publication No. H 11-21452

[Patent Reference 4]

Unexamined Patent Publication No. H 11-21453

[0007]

[Problems to be solved by the invention]

[0008]

The present invention has been accomplished in view of the above circumstances, and an object thereof is to provide a resin composition for printed wiring board which has the same molding property and workability as those of a thermosetting resin material such as a conventional epoxy resin, etc., and having excellent humidity resistance and heat resistance, and also shows particularly excellent dielectric characteristics at high frequency band and excellent stability of dielectric characteristics against change in temperature, and a varnish, a prepreg and a metal clad laminated board using the same.

[0009]

[Means for solving the problems]

The present inventors have carried our earnest

studies, and as a result, they have found that a resin composition for printed wiring board which can ensure excellent humidity resistance, has excellent dielectric characteristics at high frequency band, is a little in drift property against temperature change in dielectric characteristics, and shows an excellent stability, by using an epoxy resin having a biphenyl structure (hereinafter referred to as an epoxy resin containing a biphenyl structure) in the molecule as at least one kind of an epoxy resin when an epoxy resin is formulated to a resin composition containing a cyanate ester resin), a monovalent phenol compound) and a polyphenylene ether resin), to accomplish the present invention. Also, the resin compositions for printed wiring board of the present invention are excellent in heat resistance under humid conditions and have high strength and elongation at a glass state region or elongation at high temperature region as compared with the case where a cyanate ester resin, etc. has used alone, or the case where a conventional epoxy resin is used in combination with the cyanate ester resin, etc., so that it is preferably used for a multi-layered printed wiring board of 10-layers or more, which is required to have crack resistance at drilling or reflow or to have severe heat resistance, to be used for laminated boards and prepreg use.

[0010]

In a cyanate ester resin composition into which the conventionally employed epoxy resin is formulated, an isocyanuric ring, an oxazolidinone ring, etc. which have higher polarity than a triazine ring are formed other than the triazine ring, so that it can be considered that dielectric characteristics (in particular, dielectric dissipation factor) are worsened. On the other hand, in the resin composition containing an epoxy resin containing a biphenyl structure of the present invention, a biphenyl group show hydrophobic property and low polarity, so that

it can be considered that bad effects caused by using an epoxy resin in combination on dielectric characteristics are reduced than the conventional one. Also, a biphenyl skeleton with a brittle structure is introduced into the cured product, so that it can be considered that a temperature-dependency of dielectric characteristics is a little due to low mobility of a molecular chain particularly at a high temperature region. In addition, biphenyl skeleton-introduced system shows high strength and elongation at the glass state region and high elongation at a high temperature region as compared with the system in which the conventional epoxy resin is used in combination, so that a resin cured product excellent in crack resistance and extremely high heat resistance can be obtained.

[0011]

Thus, the present invention relates to a composition obtained by using (A) a cyanate ester compound having 2 or more cyanate groups in the molecule and/or a prepolymer thereof, (B) a monovalent phenol compound, (C) a polyphenylene ether resin, and (D) an epoxy resin containing at least one kind of an epoxy resin having a biphenyl structure in the molecule, and a varnish, a prepreg and a metal clad laminated board using the same.

[0012]

Also, the present invention relates to a resin composition for printed wiring board containing Components (A) to (D), and a varnish, a prepreg and a metal clad laminated board using the same.

[0013]

Moreover, the present invention relates to a resin composition for printed wiring board containing a phenol-modified cyanate ester oligomer obtained by reacting Component (A) with Component (B), Component (C) and Component (D), and a varnish, a prepreg and a metal clad laminated board using the same.

[0014]

Also, the present invention relates to a resin composition for printed wiring board containing Component (A), an epoxy/phenol-modified cyanate ester oligomer obtained by reacting Component (B) with Component (D), and Component (C), and a varnish, a prepreg and a metal clad laminated board using the same.

[0015]

[Embodiments of the invention]

In the following, the present invention is explained in more detail.

[0016]

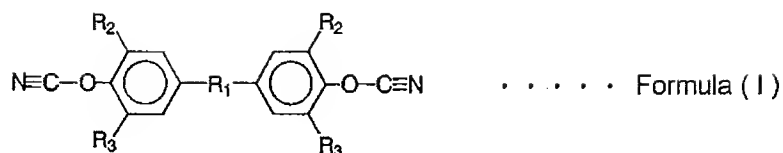
The resin composition for printed wiring board of the present invention relates to a composition obtained by using (A) a cyanate ester compound having 2 or more cyanate groups in the molecule and/or a prepolymer thereof, (B) a monovalent phenol compound, (C) a polyphenylene ether resin, and (D) an epoxy resin containing at least one kind of an epoxy resin having a biphenyl structure in the molecule. The first embodiment of the resin composition for printed wiring board according to the present invention is a resin composition for printed wiring board containing Components (A) to (D).

[0017]

(A) The cyanate ester compound having 2 or more cyanate groups in the molecule of the present invention is not specifically limited. If two or more cyanate groups are present in the molecule, it is possible to cure the composition by cross-linking. For example, there may be mentioned a cyanate ester compound represented by the formula (I):

[0018]

[Formula 10]

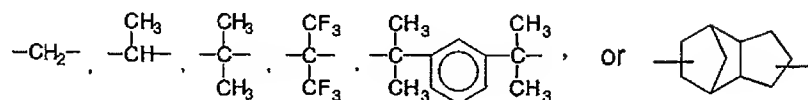


[0019]

(wherein R₁ represents

[0020]

[Formula 11]



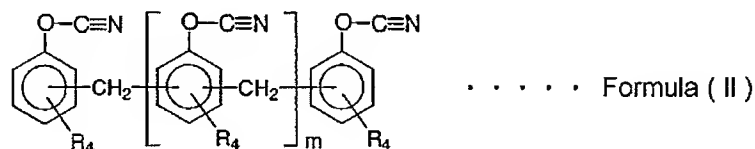
[0021]

R₂ and R₃ each represent a hydrogen atom or an alkyl group having 1 to 4 carbon atoms, and each may be the same or different from each other)

and a cyanate ester compound represented by the formula (II):

[0022]

[Formula 12]



[0023]

(wherein R₄ represents a hydrogen atom or an alkyl group having 1 to 4 carbon atoms, m represents an integer of 1 to 7.)

[0024]

A prepolymer of the cyanate ester compound having 2 or more cyanate groups in the molecule of Component (A) of the present invention is not specifically limited. Here, the prepolymer means a cyanate ester oligomer in which cyanate ester compounds form a triazine ring by cyclization

reaction, and may be mentioned mainly 3, 5, 7, 9 and 11-oligomers of the cyanate ester compounds. In the prepolymer, a conversion rate of the cyanate group is not specifically limited, and a prepolymer in which they are converted generally in the range of 20 to 70% is preferably used, more preferably 30 to 65%. For example, there may be mentioned a prepolymer of the cyanate ester compound represented by the formula (I) or (II).

[0025]

Component (A) of the present invention is preferably a compound represented by the formula (I) and its prepolymer in the point of dielectric characteristics, and preferably a compound represented by the formula (II) having high T_g in the point of heat resistance. These may be selected depending on the desired characteristics, and when they are used in combination, a formulation ratio can be optionally adjusted.

[0026]

The cyanate ester compound and a prepolymer thereof represented by the formula (I) of the present invention may be mentioned 2,2-bis(4-cyanatophenyl)propane, bis(4-cyanatophenyl)ethane, bis(3,5-dimethyl-4-cyanatophenyl)-methane, 2,2-bis(4-cyanatophenyl)-1,1,1,3,3,3-hexafluoropropane, α,α' -bis(4-cyanatophenyl)-m-diisopropylbenzene, a cyanate ester compound of a phenol-added dicyclopentadiene polymer and a prepolymer thereof, and the cyanate ester compound represented by the formula (II) and a prepolymer thereof may be mentioned a phenol novolac type cyanate ester compound and a cresol novolac type cyanate ester compound and a prepolymer thereof, etc. They may be used alone or in combination of two or more kinds.

[0027]

(B) The monovalent phenol compound of the present invention is not specifically limited. By formulating Component (B), a triazine ring can be effectively formed at the time of curing, and the cyanate group remained in the

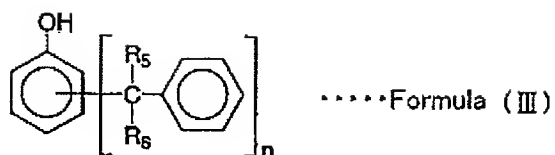
cured product as an unreacted material is imidecarbonated to reduce its polarity, whereby dielectric constant or dielectric dissipation factor of the cured product can be lowered. In the curing reaction of a usual cyanate ester sole system, a triazine ring always has three cyanate groups, so that accompanying with the progress of the reaction, the triazine ring becomes necessarily cross-linking point. In curing of the resin composition according to the present invention, one or two molecules of Component (B) is/are incorporated into a constitutional component of the triazine ring, so that a cyanate group extending from the triazine ring becomes one or two whereby the triazine ring is not necessarily a cross-linking point. That is, as compared with the cured product of the cyanate ester sole system, it becomes a cured product having characteristics that a molecular weight between the cross-linking points is large and a cross-linking density is small. According to this curing reaction, the molecular weight between the cross-linking points is large so that mobility of the molecular chain is increased and reactivity of the cyanate group becomes high, and even when the reaction proceeds, increase in viscosity is little. Accordingly, a time until the reaction system lost fluidity becomes long whereby the triazine ring can be effectively formed. As a result, an amount of the cyanate group remained in the cured produce is lowered, so that dielectric characteristics become good. Component (B) is not particularly limited so long as it is suitable for this object, and a monovalent phenol compound which is monofunctional, relatively low molecular weight and miscibility with the cyanate ester resin is good is considered to be suitable.

[0028]

As Component (B), there may be mentioned, for example, a monovalent phenol compound represented by the formula (III):

[0029]

[Formula 13]



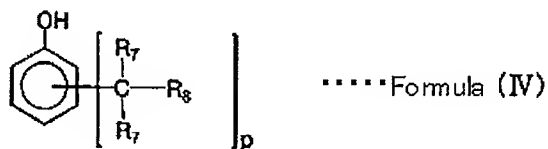
[0030]

(wherein R_5 and R_6 each represent a hydrogen atom or a methyl group, each may be the same or different from each other, and n represents an integer of 1 to 3)

and a monovalent phenol compound represented by the formula (IV):

[0031]

[Formula 14]

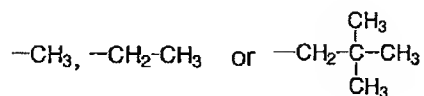


[0032]

(wherein R_7 represents a hydrogen atom or a methyl group, R_8 represents

[0033]

[Formula 15]



[0034]

and p represents an integer of 1 or 2.)

[0035]

As the monovalent phenol compound represented by the formula (III), there may be mentioned p-(α -cumyl)phenol, mono-, di- or tri-(α -methylbenzyl)phenol. As the monovalent phenol compound represented by the formula (VI), there may be mentioned p-tert-butylphenol, 2,4- or 2,6-di-tert-butylphenol, p-tert-amylphenol and p-tert-octylphenol. These monovalent phenol compounds may be used alone or in combination of two or more kinds.

[0036]

(C) The polyphenylene ether resin of the present invention is not specifically limited. By formulating Component (C), improvement in dielectric characteristics can be further realized. Incidentally, the cyanate ester compound and the polyphenylene ether resin are inherently immiscible system and a uniform resin can be difficultly obtained, but in the present invention, it is possible to obtain a uniform resin by the so-called "formation of semi-IPN" by presenting a polyphenylene ether at the time of curing, and at the time of modification of Component (A) by Component (B). Compatibilization (uniformerization) at this time is not to form chemical bond(s) by the respective component(s), but a curing component becomes oligomerization in which a curing component(s) is/are involved in molecular chain(s) of the polymer of the polyphenylene ether resin, and finally existing as resins compatibilized with each other. As described above, in the present invention, Component (B) is formulated in Component (A), so that the molecular weight between the cross-linking points of the cured product becomes large, and the curing component and the polyphenylene ether are easily involved with each other, whereby improvement in solubility can be found out.

[0037]

Component (C) of the present invention may be mentioned, for example, poly(2,6-dimethyl-1,4-phenylene)-ether, an alloy polymer comprising poly(2,6-dimethyl-1,4-phenylene)ether and an alloy polymer comprising polysty-

rene, poly(2,6-dimethyl-1,4-phenylene)ether and a styrene-butadiene copolymer and the like. When an alloy polymer comprising poly(2,6-dimethyl-1,4-phenylene)ether and polystyrene, and an alloy polymer comprising poly(2,6-dimethyl-1,4-phenylene)ether and styrene-butadiene copolymer and the like is used, and a polymer containing 50% or more of the poly(2,6-dimethyl-1,4-phenylene)ether component is preferred.

[0038]

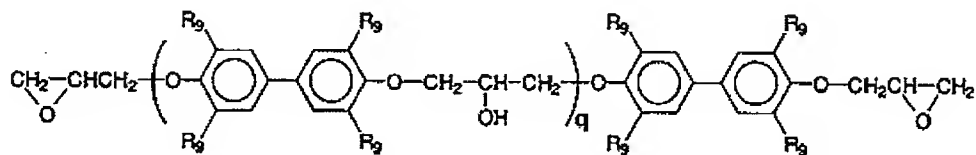
The epoxy resin containing at least one kind of (D) an epoxy resin containing a biphenyl structure of the present invention is not specifically limited so long as it contains an epoxy resin containing a biphenyl structure as an essential component. For example, it may be either of an epoxy resin comprising one kind of an epoxy resin containing a biphenyl structure, an epoxy resin comprising two or more kinds of epoxy resins containing a biphenyl structure, or an epoxy resin comprising a combination of at least one kind of an epoxy resin containing a biphenyl structure and at least one kind of another epoxy resin. Incidentally, an epoxy resin having a biphenyl skeleton and other skeleton is classified into an epoxy resin containing a biphenyl structure in the present invention.

[0039]

The epoxy resin containing a biphenyl structure is not specifically limited, and may be mentioned, for example, an epoxy resin represented by the formula (V):

[0040]

[Formula 16]



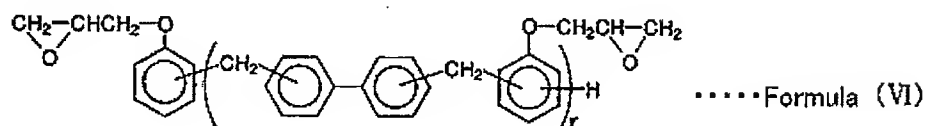
.....Formula (V)

[0041]

(wherein R₉ each represent a hydrogen atom or a methyl group, q represents an integer of 0 to 6) and an epoxy resin represented by the formula (VI):

[0042]

[Formula 17]



[0043]

(wherein r represents an integer of 1 to 5.)

[0044]

In the epoxy resin represented by the formula (V), q is preferably 0 to 5, more preferably 0 to 3, more preferably 0 or 1, particularly preferably 0. Also, a mixture of epoxy resins of the formula (V) in which values of q are different from each other may be used. There may be specifically mentioned biphenol diglycidyl ether, and 3,3',5,5'-tetramethylbiphenol diglycidyl ether, etc., and as the epoxy resin represented by the formula (VI), there may be mentioned a biphenyl aralkylene novolac type epoxy resin. Also, a mixture of epoxy resins of the formula (VI) in which values of r are different from each other may be used. These epoxy resins containing a biphenyl structure may be used alone or in combination of two or more kinds.

[0045]

When Component (D) of the present invention is a component in which at least one of the epoxy resin containing a biphenyl structure and at least one of an epoxy resin other than the epoxy resin containing a biphenyl structure are combined, the other epoxy resin is not specifically limited, and there may be mentioned, for example, a bisphenol A type epoxy resin, a bisphenol F type epoxy resin, a brominated bisphenol A type epoxy resin, a

phenol novolac type epoxy resin, a cresol novolac type epoxy resin, a brominated phenol novolac type epoxy resin, a bisphenol A novolac type epoxy resin, a naphthalene skeleton-containing epoxy resin, an aralkylene skeleton-containing epoxy resin, a phenol salicylic aldehyde novolac type epoxy resin, a lower alkyl group-substituted phenol salicylic aldehyde novolac type epoxy resin and a dicyclopentadiene skeleton-containing epoxy resin, etc. Of these, a phenol novolac type epoxy resin and a cresol novolac type epoxy resin are suitably used in combination in view of high glass transition temperature (Tg) and cost, and for the purpose of providing flame resistance, a brominated bisphenol A type epoxy resin and a brominated phenol novolac type epoxy resin, etc. are suitably used in combination.

[0046]

In the resin composition of the first embodiment, preferred ranges of formulation amounts of Components (A) to (D) are as mentioned below.

[0047]

A formulation amount of Component (B) of the present invention is preferably 2 to 60 parts by weight based on 100 parts by weight of Component (A), more preferably 3 to 45 parts by weight, particularly preferably 4 to 30 parts by weight. When Component (B) is formulated in this range, in particular, good dielectric characteristics that dielectric dissipation factor at high frequency band region is sufficiently low can be obtained, and, good heat resistance at humidity conditions can be obtained.

[0048]

A formulation amount of Component (C) of the present invention is preferably 5 to 300 parts by weight based on 100 parts by weight of Component (A), more preferably 10 to 200 parts by weight, particularly preferably 15 to 100 parts by weight. When a formulation amount of Component (C) is in this range, sufficient dielectric characteristics

can be obtained, and a melt viscosity of the resin is suitable so that fluidity is sufficient and molding property is good, and there is a tendency that a reactivity of Component (A) is also good.

[0049]

A formulation amount of Component (D) of the present invention is preferably 10 to 250 parts by weight based on 100 parts by weight of Component (A) in the points of humidity resistance under severe conditions, heat resistance or strength at the time of humid conditions and improvement in elongation, etc., and dielectric characteristics at high frequency band region. A formulation amount of Component (D) is more preferably 10 to 150 parts by weight, particularly preferably 10 to 100 parts by weight. In the point of dielectric characteristics, a ratio of the epoxy resin containing a biphenyl structure in Component (D) is preferably 50 % by weight or more, more preferably 70 % by weight or more, particularly preferably 100 % by weight, i.e., all the Component (D) is the epoxy resin containing a biphenyl structure.

[0050]

To the resin composition of the first embodiment, (E) a flame retardant can be added. Component (E) is not specifically limited, and a flame retardant having any reactivity with a cyanate group is preferred. Here, the terms of having no reactivity with a cyanate group mean that a flame retardant is added to a resin composition for printed wiring board, the flame retardant does not react with the cyanate group of a cyanate ester compound even when they are mixed in the range of 300°C or lower, and contained in a resin composition for printed wiring board as such in the state of a dispersion or dissolution. In this reaction, a reaction(s) of the flame retardant at the time of heating and burning the resin composition is/are not included. In general, production and use of the resin composition for printed wiring board, and the varnish,

prepreg, metal clad laminated board and printed wiring board, etc. using the same are carried out within the range of 300°C or lower.

[0051]

As such a flame retardant, there may be mentioned a flame retardant substantially having no alcoholic hydroxyl group, phenolic hydroxyl group, amino group or maleimide group, which are known as a group which reacts with a cyanate group. Here, the terms "substantially having no" include the case where the flame retardant never contains these groups and the case where even when the flame retardant contains these groups but there is no interaction with the cyanate ester compound (for example, it is the case where the flame retardant is a polymer having a large molecular weight, and these groups exist at the end of the molecule, so that there is no effect as a functional group).

[0052]

Incidentally, with regard to a flame retardant classified into an epoxy resin including a brominated epoxy resin such as a brominated bisphenol A type epoxy resin and a brominated phenol novolac type epoxy resin, etc., it is classified in Component (D) in the present invention. Provided that, a flame retardant classified into an epoxy compound having a glycidyl group reacts with a cyanate ester compound, and can be inserted into a triazine ring or an isocyanuric ring, but if it is a suitable formulation amount, effects on dielectric characteristics are considered to be little so that it may be used as a flame retardant.

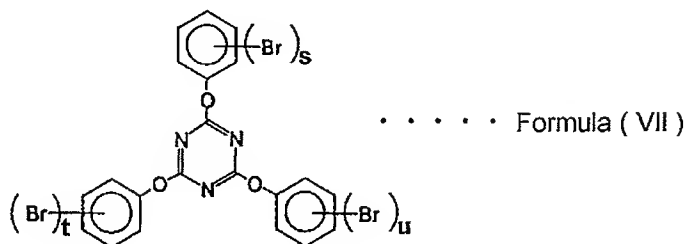
[0053]

Specific flame retardant may be mentioned, for example, 1,2-dibromo-4-(1,2-dibromoethyl)cyclohexane, tetrabromocyclooctane, hexabromocyclododecane, bis(tri-bromophenoxy)ethane, a brominated polyphenylene ether, brominated polystyrene, and a brominated triphenylcyanurate

flame retardant represented by the formula (VII):

[0054]

[Formula 18]



[0055]

(wherein s, t and u each represent an integer of 1 to 5, and each may be the same value or different from each other)

and the like, and in the point of dielectric characteristics, 1,2-dibromo-4-(1,2-dibromoethyl)cyclohexane, tetrabromocyclooctane, hexabromocyclododecane and 2,4,6-tris-(tribromophenoxy)-1,3,5-triazine are preferred.

[0056]

A formulation amount of the flame retardant is preferably 5 to 100 parts by weight based on the total amount of Components (A) to (D) as 100 parts by weight, more preferably 5 to 80 parts by weight, particularly preferably 5 to 60 parts by weight. By formulating it in this range, flame resistance is sufficient in the resin composition, and it is preferred for heat resistance of the cured product.

[0057]

To the resin composition of the first embodiment, (F) an antioxidant may be added. By formulating the antioxidant, when the resin composition for printed wiring board is cured and fabricated to laminated boards, etc., occurrence of metal migration can be controlled and further improvement in insulation reliability can be established.

[0058]

Component (F) of the present invention is not

specifically limited, and a phenol type antioxidant and a sulfur type antioxidant are preferably used. Incidentally, it has been well known for those skilled in the art that the antioxidant is classified from the skeletal structure, and for example, in "Antioxidant Handbook" pp. 12 to 17 (1976), specific antioxidants are exemplified as a classification of "a phenol type antioxidant" and "a sulfur type antioxidant".

[0059]

Specific examples of the phenol type antioxidant may include a monophenol type one such as pyrogallol, butylated hydroxyanisole, 2,6-di-tert-butyl-4-methylphenol, etc., a bisphenol type one such as 2,2'-methylene-bis-(4-methyl-6-tert-butylphenol), 4,4'-butylidenebis(3-bis-6-tert-butylphenol), etc. and a polymer type phenol series one such as 1,3,5-trimethyl-2,4,6-tris(3,5-di-tert-butyl-4-hydroxybenzyl)benzene, tetrakis-[methylene-3-(3'-5'-di-tert-butyl-4'-hydroxyphenyl)propionate]methane, etc. Specific examples of the sulfur series antioxidant may include dilaurylthiodipropionate, distearylthiodipropionate, etc. These antioxidants may be used alone or in admixture of two or more.

[0060]

A formulation amount of Component (F) of the present invention is preferably 0.1 to 20 parts by weight based on the total amount of Components (A) to (D) as 100 parts by weight, more preferably 0.1 to 10 parts by weight, particularly preferably 0.1 to 5 parts by weight. If it is formulated in this range, a preferred resin composition for printed wiring board can be obtained in the points of improved effects for insulation reliability of the cured product or laminated boards.

[0061]

In the resin composition of the first embodiment, a metal series catalyst is preferably formulated. The metal series catalyst acts as a promoting agent of a self-

polymerization reaction of Component (A), in the reaction of Component (A) and Component (B), and in the reaction of Component (A) and Component (D), and as a curing accelerator at the time of producing laminated boards, and there may be mentioned, for example, a metal salt or chelate complex of a As the metal, there may be mentioned, for example, copper, cobalt, manganese, iron, nickel, zinc, etc., and as the salt thereof, there may be mentioned, for example, a metal salt such as a carboxylate (preferably 2-ethylhexanoate, naphthenate), etc., and as the chelate complex, there may be mentioned, for example, an acetylacetone complex. These metal series catalysts may be used alone or in combination of two or more kinds. Also, the promoting agent in the self-polymerization reaction of Component (A), and in the reaction of Component (A) and Component (C), and the curing accelerator at the time of producing laminated boards may be the same or different metal series catalyst from each other.

[0062]

An amount of the metal series catalyst is preferably 1 to 300 ppm based on the amount of Component (A) in terms of weight, more preferably 1 to 200 ppm, particularly preferably 2 to 150 ppm. When a metal series catalyst is formulated in this range, reactivity and curing property is sufficient, and curing rate is suitable. Addition of the metal series catalyst may be carried out at once or may be divided into a plural number of times.

[0063]

In the resin composition of the first embodiment, a compound having a catalytic function which promotes the reaction of the glycidyl group of Component (D) can be formulated. More specifically, there may be mentioned, an alkali metal compound, an alkaline earth metal compound, an imidazole compound, an organic phosphorus compound, a secondary amine, a tertiary amine, a quaternary ammonium salt, etc. These compounds may be used alone or in

combination of two or more kinds.

[0064]

In the resin composition of the first embodiment, an additive such as a filler, etc. may be formulated within a range which does not worsen characteristics such as dielectric characteristics or heat resistance, etc. of the cured product.

[0065]

The filler to be formulated depending on necessity is not specifically limited, it is usually an inorganic filler, and there may be mentioned, for example, alumina, titanium oxide, mica, silica, beryllia, barium titanate, potassium titanate, strontium titanate, calcium titanate, aluminum carbonate, aluminum hydroxide, aluminum silicate, potassium carbonate, calcium silicate, magnesium silicate, silicon nitride, boron nitride, clay (calcined clay, etc.), talc, aluminum borate, aluminum borate, silicon carbide, etc. These inorganic fillers may be used alone or in combination of two or more kinds. Also, a shape, a particle size, etc. of the inorganic filler are not specifically limited, and it is usually having a particle size of 0.01 to 50 μm , preferably those having 0.1 to 15 μm . Moreover, a formulation amount of these inorganic filler is not specifically limited, and is preferably 1 to 1000 parts by weight based on 100 parts by weight of Component (A), Component (B) and Component (C) (when they are formulated) in total, more preferably 1 to 800 parts by weight.

[0066]

The resin composition of the first embodiment can be produced by formulating and mixing Components (A) to (D) and other additives by the conventionally known method.

[0067]

The second embodiment of the present invention is a resin composition for printed wiring board characterized in containing a phenol-modified cyanate ester oligomer

obtained by reacting (A) a cyanate ester compound having 2 or more cyanate groups in the molecule and/or a prepolymer with (B) a monovalent phenol compound, and (C) a polyphenylene ether resin and (D) an epoxy resin containing at least one kind of an epoxy resin containing a biphenyl structure.

[0068]

The phenol-modified cyanate oligomer becomes a mixed oligomer comprising, for example, a cyanate ester oligomer in which Component (A) solely forms a triazine ring by cyclization reaction, an imidecarbonated-modified oligomer in which a phenolic hydroxyl group of Component (B) is added to a cyanate group of Component (A), and/or a modified oligomer (in this case, among the three chains extended from the triazine ring, one or two thereof is/are replaced by the molecule derived from Component (B)) in which one or two of Component (B) is introduced into the structure of a triazine ring formed by sole cyclization of Component (A).

[0069]

With regard to examples of Component (A) and Component (B) to be used for formation of the phenol-modified cyanate ester oligomer, and examples of Component (C) and Component (D), descriptions of Components (A) to (D) with regard to the first embodiment can be applied to.

[0070]

Component (B) to be used for formation of the phenol-modified cyanate ester oligomer is preferably 2 to 60 parts by weight based on 100 parts by weight of Component (A), more preferably 3 to 45 parts by weight, particularly preferably 4 to 30 parts by weight.

[0071]

Component (B) to be used for formation of the phenol-modified cyanate ester oligomer is preferably 2 to 60 parts by weight based on 100 parts by weight of Component (A), more preferably 3 to 45 parts by weight, particularly

preferably 4 to 30 parts by weight.

[0072]

The resin composition of the second embodiment may further contain Component (B), and in such a case, a sum of said Component (B), and Component (B) to be used for formation of the phenol-modified cyanate ester oligomer is preferably within the range of 2 to 60 parts by weight based on 100 parts by weight of Component (A). For example, Component (B) is previously reacted in the range of 0.4 parts by weight or more and less than 60 parts by weight based on 100 parts by weight of Component (A) to prepare a phenol-modified cyanate ester oligomer, and then, Component (B) is additionally formulated so that a sum of the amount thereof with that of Component (B) to be used for formation of the phenol-modified cyanate ester oligomer becomes within the range of 2 to 60 parts by weight. When Component (B) is previously reacted in the range of 2 to 60 parts by weight based on 100 parts by weight of Component (A) to obtain a phenol-modified cyanate ester oligomer, additional Component (B) may not be formulated, or may be formulated so that a sum of the amount thereof with that of Component (B) to be used for formation of the phenol-modified cyanate ester oligomer becomes within the range of 2 to 60 parts by weight. Incidentally, in the above-mentioned case, Component (B) to be used for formation of the phenol-modified cyanate ester oligomer, and additional Component (B) may be the same or different from each other, and each may be used in combination of two or more kinds in admixture.

[0073]

The phenol-modified cyanate ester oligomer can be prepared by, for example, dissolving Component (A) and Component (B) in a solvent such as toluene, xylene, mesitylene, etc., and heating at 70 to 120°C for 0.5 to 10 hours. At this time, the metal series catalyst described in the first embodiment may be added. These metal series

catalysts are to promote modification by phenol. Incidentally, the phenol-modified cyanate ester oligomer may be made, for example, a conversion of the initial cyanate group of 20 to 70%, preferably within the range of 30 to 65%.

[0074]

Incidentally, formation of the phenol-modified cyanate ester oligomer is preferably carried out in the presence of Component (C). Component (C) can be made a melting product by heating or a solution in which it is dissolved in a solvent. According to this, it is possible to establish the so-called "semi-IPN" in which a phenol-modified cyanate ester oligomer and Component (C) are uniformly compatibilized. When it is made a solution, as a solvent, there may be mentioned a solvent such as alcohols such as methanol, ethanol, ethylene glycol, ethylene glycol monomethyl ether, etc., ketones such as acetone, methyl ethyl ketone, methyl isobutyl ketone, cyclohexanone, etc., aromatic hydrocarbons such as toluene, xylene, mesitylene, etc., esters such as methoxyethyl acetate, ethoxyethyl acetate, butoxyethyl acetate, ethyl acetate, etc., amides such as N-methylformamide, N,N-dimethylformamide, N,N-dimethylacetamide, N-methylpyrrolidone, etc. In particular, aromatic hydrocarbons such as toluene, xylene, mesitylene, etc. are more preferred.

[0075]

In the resin composition of the second embodiment, (E) a flame retardant may be formulated similarly as in the resin composition of the first embodiment. With regard to examples of Component (E), and a formulation amount thereof, descriptions with regard to the resin composition of the first embodiment can be applied to. Also, in the resin composition of the second embodiment, (F) an anti-oxidant may be formulated. With regard to examples of Component (F), and a formulation amount thereof, descriptions with regard to the resin composition of the

first embodiment can be applied to. In the formulation amounts of the flame retardant and the antioxidant, in the sum of Components (A) to (D), Component (A) and Component (B) which are starting materials of the phenol-modified cyanate ester oligomer are to be contained.

[0076]

In the resin composition of the second embodiment, a metal series catalyst, a compound having a catalytic function which promotes the reaction of the glycidyl group of an epoxy resin, a filler and the other additives, etc. can be formulated, and with regard to examples thereof, formulation amounts, and the method of producing the resin composition, descriptions with regard to the first embodiment can be applied to. Incidentally, in the formulation amount of the filler, in the sum of Components (A) to (D), Component (A) and Component (B) which are starting materials of the phenol-modified cyanate ester oligomer are to be contained.

[0077]

The resin composition of the third embodiment is a resin composition for printed wiring board characterized in containing an epoxy/phenol-modified cyanate ester oligomer obtained by reacting (A) a cyanate ester compound having 2 or more cyanate groups in the molecule and/or a prepolymer, (B) a monovalent phenol compound and (D) an epoxy resin containing at least one kind of an epoxy resin containing a biphenyl structure, and Component (C).

[0078]

With regard to examples of Component (A), Component (B) and Component (D) to be used for formation of the epoxy/phenol-modified cyanate ester oligomer, descriptions of Component (A), Component (B) and Component (D) with regard to the first embodiment can be applied to. For formation of the epoxy/phenol-modified cyanate ester oligomer, Component (B) is preferably 2 to 60 parts by weight based on 100 parts by weight of Component (A), more

preferably 3 to 45 parts by weight, particularly preferably 4 to 30 parts by weight. Component (D) is preferably 10 to 250 parts by weight based on 100 parts by weight of Component (A), more preferably 10 to 150 parts by weight, particularly preferably 10 to 100 parts by weight. In the point of dielectric characteristics, a ratio of the epoxy resin containing a biphenyl structure in Component (D) is preferably 50 % by weight or more, more preferably 70 % by weight or more, particularly preferably 100 % by weight, i.e., all the Component (D) is the epoxy resin containing a biphenyl structure. With regard to Component (C), descriptions of Component (C) with regard to the first embodiment can be applied to. A formulation amount of Component (C) is preferably 5 to 300 parts by weight based on 100 parts by weight of Component (A) which is used for formation of the epoxy/phenol-modified cyanate ester oligomer, more preferably 10 to 200 parts by weight, particularly preferably 15 to 100 parts by weight.

[0079]

The epoxy/phenol-modified cyanate ester oligomer may further contain Component (B), and in such a case, a sum of said Component (B), and Component (B) to be used for formation of the epoxy/phenol-modified cyanate ester oligomer is preferably within the range of 2 to 60 parts by weight based on 100 parts by weight of Component (A). For example, 0.4 parts by weight or more and less than 60 parts by weight of Component (B), and 10 to 250 parts by weight of Component (D) are previously reacted based on 100 parts by weight of Component (A) to prepare an epoxy/phenol-modified cyanate ester oligomer, and then, Component (B) is additionally formulated so that a sum of the amount thereof with that of Component (B) to be used for formation of the an epoxy/phenol-modified cyanate ester oligomer becomes within the range of 2 to 60 parts by weight. When 2 to 60 parts by weight of Component (B), and 10 to 250 parts by weight of Component (D) based on 100 parts by weight of

Component (A) are provisionally reacted to prepare the epoxy/phenol-modified cyanate ester oligomer, additional Component (B) may not be formulated, or may be formulated so that a sum of the amount thereof with that of Component (B) to be used for formation of the epoxy/phenol-modified cyanate ester oligomer becomes within the range of 2 to 60 parts by weight. Incidentally, in the above-mentioned case, Component (B) to be used for formation of the epoxy/phenol-modified cyanate ester oligomer, and additional Component (B) may be the same or different from each other, and each may be used in combination of two or more kinds in admixture.

[0080]

The phenol-modified cyanate ester oligomer can be prepared, for example, Component (A), and Component (B) and Component (D) are mixed in a solvent such as toluene, xylene, mesitylene, etc., and heating to 70 to 120°C for 0.5 to 10 hours. At this time, the metal series catalyst described in the first embodiment may be added to the mixture. Incidentally, the phenol-modified cyanate ester oligomer can be made, for example, a conversion of the initial cyanate group of 20 to 70%, preferably in the range of 30 to 65%.

[0081]

Incidentally, formation of the epoxy phenol-modified cyanate ester oligomer is preferably carried out in the presence of Component (C). Epoxy/phenol-modification can be carried out in a melting product of Component (C) under heating, or in a solution dissolved in a solvent. The solvent can be applied to those as mentioned in the second embodiment. According to the above, the so-called "semi-IPN" in which the epoxy/phenol-modified cyanate ester oligomer and Component (C) are uniformly compatibilized can be obtained.

[0082]

In the resin composition of the third embodiment, (E)

a flame retardant can be formulated similarly as in the resin composition of the first embodiment. With regard to examples of Component (E), and a formulation amount thereof, descriptions with regard to the resin composition of the first embodiment can be applied to. Also, in the resin composition of the third embodiment, (F) an antioxidant can be formulated. With regard to examples of Component (F), and a formulation amount thereof, descriptions with regard to the resin composition of the first embodiment can be applied to. In the formulation amounts of the flame retardant and the antioxidant, in the sum of Components (A) to (D), Component (A), Component (B) and Component (D) which are starting materials of the phenol-modified cyanate ester oligomer are to be contained.

[0083]

In the resin composition of the third embodiment, a metal series catalyst, a compound having a catalytic function which promotes the reaction of the glycidyl group of an epoxy resin, a filler and the other additives, etc. can be formulated, and with regard to examples thereof, formulation amounts, and the method of producing the resin composition, descriptions with regard to the first embodiment can be applied to. Incidentally, in the formulation amount of the filler, in the sum of Components (A) to (D), Component (A), Component (B) and Component (D) which are starting materials of the epoxy/phenol-modified cyanate ester oligomer are to be contained.

[0084]

By using the resin composition for printed wiring board of the first to the third embodiment according to the present invention, a prepreg for a printed wiring board or a metal clad laminated board can be produced according to the conventionally known method. For example, the resin composition for printed wiring board of the present invention is as such, or in a form of a varnish in which they are dissolved or dispersed in a solvent, impregnated

into a substrate such as a glass cloth, etc., and then, drying in a drying oven usually at a temperature of 80 to 200°C (provided that, when a solvent is used, it is set at a temperature capable of volatilizing the solvent or higher), preferably at 100 to 180°C, for 3 to 30 minutes, preferably 3 to 15 minutes to obtain a prepreg. Then, this prepreg is laminated with a plural number of sheets, a metal foil(s) is/are provided on one surface or both surfaces thereof, and molding under heating to produce a both surface or one surface metal clad laminated board.

[0085]

Incidentally, the solvent to be used in the above-mentioned varnishing treatment is not specifically limited, and there may be mentioned a solvent, for example, alcohols such as methanol, ethanol, ethylene glycol, ethylene glycol monomethyl ether, etc., ketones such as acetone, methyl ethyl ketone, methyl isobutyl ketone, cyclohexanone, etc., aromatic hydrocarbons such as toluene, xylene, mesitylene, etc., esters such as methoxyethyl acetate, ethoxyethyl acetate, butoxyethyl acetate, ethyl acetate, etc., amides such as N-methylformamide, N,N-dimethylformamide, N,N-dimethylacetamide, N-methylpyrrolidone, etc. In particular, aromatic hydrocarbons such as toluene, xylene, mesitylene, etc. are more preferred. They may be used alone or in combination of two or more kinds. When it is used as a mixed solvent, if the above-mentioned aromatic hydrocarbons and ketones such as acetone, methyl ethyl ketone, methyl isobutyl ketone, cyclohexanone, etc. are used in combination, a viscosity of the varnish can be lowered, so that it is preferred in the point of obtaining a high-concentration varnish. A formulation amount of the ketone series solvents is preferably used in an amount of 30 to 300 parts by weight based on 100 parts by weight of the aromatic hydrocarbon series solvents, more preferably 30 to 250 parts by weight, further preferably 30 to 220 parts by weight.

[0086]

The resin composition for printed wiring board of the present invention, and the varnish, prepreg and metal clad laminated board using the same can be used for a printed wiring board which can be used for information communication-related devices (parts such as a filter, VCO, etc. included in mobile communication devices, or signal processor, power amplifier and antenna which constitutes wireless base station equipments, or a high speed computer which exceeds a clock frequency of 1 GHz such as a server, router and microprocessor, etc.) which are required to have higher frequencies and high speed in signals.

[0087]

[EXAMPLE]

In the following, the present invention is explained in more detail by referring to specific examples, but the present invention is not limited by these.

[0088]

[Examples 1 to 5, Comparative examples 1 to 6]

Resin varnishes for metal clad laminated boards were produced according to the formulation amounts as shown in Table 1.

Example 1

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 300 g of toluene and 175 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of 2,2-bis(4-cyanatophenyl)propane (Arocy B-10, available from Ciba Geigy AG) and 32 g of p-tert-butylphenol (available from Kanto Kagaku) were charged therein, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.13 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a reaction promoter was formulated in the mixture, and the mixture was reacted for about 3 hours to synthesize

a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 530 g of methyl ethyl ketone and 308 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (a tetramethylbiphenyl type epoxy resin, YX-4000, available from Japan Epoxy Resin Co.) were formulated in the mixture under stirring. Then, after confirmation of dissolution, the mixture was cooled to room temperature; 0.1 g of zinc naphthenate was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0089]

Example 2

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 275 g of toluene and 100 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of bis(3,5-dimethyl-4-cyanatophenyl)methane (Arocy M-10, available from Ciba Geigy AG) and 47 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.25 g of cobalt naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a reaction promoter was formulated in the mixture, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 480 g of methyl ethyl ketone and 280 g of a mixed biphenyl type epoxy resin (YL-612H, available from Japan Epoxy Resin Co.) comprising a biphenol diglycidyl ether and of 3,3',5,5'-tetramethylbiphenol diglycidyl ether were formulated in the mixture under stirring. Then, after confirmation of

dissolution, the mixture was cooled to room temperature; 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0090]

Example 3

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 285 g of toluene and 150 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of 2,2-bis(4-cyanatophenyl)propane (Arocy B-10, available from Ciba Geigy AG) and 15 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.16 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a reaction promoter was formulated in the mixture, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 520 g of methyl ethyl ketone and 273 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 46 g of p-(α -cumyl)phenol and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0091]

Example 4

In a 3-liter four-necked separable flask equipped

with a thermometer, a condenser and a stirring device, 270 g of toluene and 180 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of α,α' -bis(4-cyanatophenyl)-m-diisopropylbenzene (RTX-366, available from Ciba Geigy AG) and 4 g of p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.14 g of iron naphthenate (available from Kanto Kagaku) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 535 g of methyl ethyl ketone and 330 g of a biphenyl aralkylene novolac type epoxy resin (NC-3000S-H, available from Nippon Kayaku Co., Ltd.) were formulated in the mixture under stirring. Then, after confirmation of dissolution, the mixture was cooled to room temperature, 23 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0092]

Example 5

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 290 g of toluene and 225 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of a phenol novolac type cyanate ester resin (REX-371, available from Ciba Geigy AG) and 13 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo

Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.15 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 2 hours under heating to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 500 g of methyl ethyl ketone, 148 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) and 90 g of a cresol novolac type epoxy resin (ESCN-190-3, available from Sumitomo Chemical Co., Ltd.) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 45 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a curing accelerator was formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0093]

Comparative Example 1

In the same manner as in Example 1 except for removing 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000), and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 1 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0094]

Comparative Example 2

In the same manner as in Example 1 except for formulating a bisphenol A type epoxy resin (DER-331L, available from Dow Chemical) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with a formulation

amount as shown in Table 3, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 1 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0095]

Comparative Example 3

In the same manner as in Example 1 except for formulating p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of p-tert-butylphenol, and a phenol novolac type epoxy resin (N-770, available from Dainippon Ink and Chemicals Incorporated) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with formulation amounts as shown in Table 3, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 1 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0096]

Comparative Example 4

In the same manner as in Example 2 except for formulating a bisphenol A novolac type epoxy resin (N-865, available from Dainippon Ink and Chemicals Incorporated) in place of a mixed biphenyl type epoxy resin (YL-612H) comprising biphenol diglycidyl ether and 3,3',5,5'-tetramethylbiphenol diglycidyl ether, and zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) in place of cobalt naphthenate (available from Wako Pure Chemical Industries, Ltd.) with formulation amounts as shown in Table 3, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 2 to prepare a resin varnish having the non-volatile concentration of about 55% by

weight.

[0097]

Comparative Example 5

In the same manner as in Example 1 except for formulating p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) in place of p-tert-butylphenol, and a phenol salicylic aldehyde novolac type epoxy resin (EPPN-502H, available from Nippon Kayaku Co., Ltd.) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with formulation amounts as shown in Table 3, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 1 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0098]

Comparative Example 6

In the same manner as in Comparative Example 4 except for formulating p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of p-tert-butylphenol, and a dicyclopentadiene skeleton-containing epoxy resin (HP-7200, available from Dainippon Ink and Chemicals Incorporated) in place of a bisphenol A novolac type epoxy resin (N-685) with formulation amounts as shown in Table 3, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Comparative Example 4 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0099]

[Table 1]

Table 1

Item	Example					Comparative example					
	1	2	3	4	5	1	2	3	4	5	6
Cyanate ester compound	500	-	500	-	-	500	500	500	-	500	-
	-	500	-	-	-	-	-	-	500	-	500
	-	-	-	450	-	-	-	-	-	-	-
	-	-	-	-	450	-	-	-	-	-	-
Phenol compound	32	-	-	-	-	32	32	-	-	-	-
	-	47	-	23	45	-	-	-	47	44	-
	-	-	15	-	13	-	-	-	-	-	48
	-	-	46	-	-	-	-	-	-	-	-
Polyphenylene ether resin	175	100	150	180	225	175	175	175	100	175	100
	308	-	273	-	148	-	-	-	-	-	-
	-	280	-	-	-	-	-	-	-	-	-
	-	-	-	330	-	-	-	-	-	-	-
Epoxy resin	-	-	-	-	90	-	-	-	-	-	-
	-	-	-	-	-	-	300	-	-	-	-
	-	-	-	-	-	-	-	306	-	-	-
	-	-	-	-	-	-	-	-	338	-	-
Reaction catalyst (Curing accelerator)	-	-	-	-	-	-	-	-	-	313	-
	-	-	-	-	-	-	-	-	-	-	460
	-	-	0.16	-	0.15	-	-	-	-	-	-
	-	0.25	-	-	-	-	-	-	-	-	-
Non-volatile concentration of varnish (% by weight)	-	-	-	0.14	-	-	-	-	-	-	-
	0.13	0.1	0.1	0.1	0.1	0.13	0.13	0.13	0.13	0.13	0.13
	0.1	-	-	-	-	0.1	0.1	0.1	0.1	0.1	0.1
	55	55	55	55	55	55	55	55	55	55	55

[0100]

Resin varnishes obtained in Examples 1 to 5 and

Comparative examples 1 to 6 were each impregnated into a glass cloth (E glass) with a thickness of 0.15 mm, and then, the material was dried at 160°C for 4 to 7 minutes to obtain the respective prepregs with a resin solid content of 52 % by weight. Then, four sheets of this prepregs were laminated, and copper foils with a thickness of each 18 μm were provided on the both outermost layers, and molding was carried out under the pressing conditions of 230°C for 70 minutes and 2.5MPa to prepare a both-surface copper clad laminated board. With regard to the obtained copper clad laminated board, dielectric characteristics, copper foil peeling strength, solder heat resistance, water absorption rate, bending characteristics, thermal expansion coefficient (α) and Tg (glass transition temperature) were evaluated. The evaluation results are shown in Table 2.

[0101]

Evaluating methods of characteristics of the copper clad laminated boards are as shown below.

Dielectric constant (ϵ_r) and dielectric dissipation factor ($\tan \delta$) of the copper clad laminated boards were measured by a triplate line resonator method using a vector type network analyzer. Incidentally, the measurement conditions are made a frequency: 1 GHz, measurement temperatures: room temperature (25°C) and 90°C.

Thermal expansion coefficient (α) and Tg of the copper clad laminated boards (whole surface-copper foil etched product) were measured by TMA.

Copper foil peeling strength of the copper clad laminated boards was measured according to the copper clad laminated boards test standard JIS-C-6481.

Solder heat resistance of the copper clad laminated boards (whole surface-copper foil etched product) was examined by maintaining in a pressure cooker tester (conditions: 121°C, 2.2 atm) for 1 to 5 hours, then, immersing in a melting solder at 260°C and 288°C for 20 seconds, and the appearance thereof was judged with naked

eyes. No abnormality in the table means that there is no occurrence of measling or blister.

Water absorption rate of the copper clad laminated boards (whole surface-copper foil etched product) was calculated from a weight difference between the normal state and after maintaining in a pressure cooker tester (conditions: 121°C, 2.2 atm) for 5 hours (unit: % by weight).

Bending characteristics of the copper clad laminated boards (whole surface-copper foil etched product) were obtained by measuring bending modulus at room temperature and at 200°C according to copper clad laminated boards test standard JIS-C-6481, breaking strength and breaking elongation at room temperature, and yield strength and yield elongation at 200°C.

[0102]

[Table 2]

Table 2

Item	Example					Comparative example					
	1	2	3	4	5	1	2	3	4	5	6
Copper foil peeling strength (kN/m)	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.4	1.4	1.4	1.4
ε	3.50	3.45	3.50	3.40	3.57	3.50	3.60	3.63	3.52	3.60	3.50
	3.52	3.46	3.51	3.42	3.58	3.51	3.65	3.69	3.59	3.66	3.58
t	0.0052	0.0046	0.0052	0.0040	0.0054	0.0050	0.0062	0.0067	0.0065	0.0066	0.0054
	0.0057	0.0051	0.0057	0.0045	0.0058	0.0053	0.0083	0.0085	0.0082	0.0080	0.0066
Solder heat resistance (260°C) (Number of no abnormality / Number of tested) (288°C)	PCT 1h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3
	PCT 2h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3
	PCT 3h	3/3	3/3	3/3	3/3	3/3	2/3	3/3	3/3	3/3	3/3
	PCT 4h	3/3	3/3	3/3	3/3	3/3	1/3	3/3	3/3	3/3	1/3
	PCT 5h	3/3	3/3	3/3	3/3	3/3	0/3	3/3	3/3	2/3	1/3
	PCT 1h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3
	PCT 2h	3/3	3/3	3/3	3/3	3/3	2/3	3/3	3/3	3/3	3/3
	PCT 3h	3/3	3/3	3/3	3/3	3/3	0/3	2/3	2/3	2/3	0/3
	PCT 4h	1/3	2/3	2/3	3/3	2/3	0/3	0/3	0/3	0/3	0/3
	PCT 5h	1/3	0/3	1/3	3/3	1/3	0/3	0/3	0/3	0/3	0/3
Water absorption rate (% by weight)	0.52	0.50	0.51	0.47	0.53	0.50	0.54	0.58	0.58	0.62	0.53
TMA	Tg (°C)	186	184	185	178	192	185	175	188	191	185
	α (ppm/°C, <Tg)	50	52	48	53	52	50	55	51	51	59
	α (ppm/°C, >Tg)	305	294	312	287	296	285	377	322	328	278
Bending modulus(GPa)	Room temperature: 25°C	18	18	18	17	19	18	18	18	18	18
	200°C	10	10	10	9	13	11	7	12	13	9
Breaking strength(MPa) (Yield strength)	Room temperature: 25°C	595	588	605	584	612	488	558	516	502	535
	(200°C)	(245)	(225)	(230)	(198)	(273)	(210)	(142)	(244)	(232)	(286)
Breaking elongation(%) (Yield strength)	Room temperature: 25°C	3.8	3.7	3.8	3.8	3.7	3.0	3.5	3.2	3.2	3.4
	(200°C)	(3.9)	(3.7)	(3.8)	(3.8)	(3.6)	(3.5)	(3.5)	(2.4)	(2.3)	(2.2)

[0103]

As can be clearly seen from Table 2, the laminated

boards prepared by using the varnishes of Examples 1 to 5 are excellent in dielectric characteristics (in particular, dielectric dissipation factor) at room temperature (25°C) and 1 GHz than those of the laminated boards of Comparative examples 2 to 5. Moreover, dielectric characteristics (in particular, dielectric dissipation factor) thereof at 90°C are also good, and dependency on temperature change is little as compared to those of Comparative examples 2 to 6. Also, the laminated boards of Examples show lower water absorption rates as compared to those of the laminated boards of Comparative examples 2 to 6 and give better results in solder heat resistance (in particular, 288°C) under humid conditions as compared to the laminated boards of Comparative examples 1 to 6. In addition, the laminated boards of Examples 1 to 5 have higher breaking strength and breaking elongation at room temperature (25°C) and yield elongation at high temperature (200°C) than those of the laminated boards of Comparative examples 1 to 6.

[0104]

[Examples 6 to 11, Comparative examples 7 to 12]

Resin varnishes for a metal clad laminated board were produced according to the formulation amounts as shown in Table 3.

[0105]

Example 6

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 390 g of toluene and 175 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of 2,2-bis(4-cyanatophenyl)propane (Arocy B-10, available from Ciba Geigy AG) and 32 g of p-tert-butylphenol (available from Kanto Kagaku) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.13 g of zinc naphthenate (available from Wako Pure Chemical Industries,

Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 4 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 608 g of methyl ethyl ketone, 308 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (a tetramethylbiphenyl type epoxy resin, YX-4000, available from Japan Epoxy Resin Co.) and 207 g of brominated polystyrene (PDBS-80, available from Great Lakes) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature and 0.1 g of zinc naphthenate was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0106]

Example 7

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 350 g of toluene and 100 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of bis(3,5-dimethyl-4-cyanatophenyl)-methane (Acocy M-10, available from Ciba Geigy AG) and 47 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.25 g of cobalt naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 4 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 550 g of methyl ethyl ketone, 280 g of a mixed biphenyl type

epoxy resin (YL-612H, available from Japan Epoxy Resin Co.) comprising biphenol diglycidyl ether and 3,3',5,5'-tetramethylbiphenol diglycidyl ether, and 178 g of a brominated polyphenylene ether (PO-64P, available from Great Lakes) were formulated in the mixture under stirring. Then, after confirmation of dissolution, the mixture was cooled to room temperature, and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight. [0107]

Example 8

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 357 g of toluene and 150 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of 2,2-bis(4-cyanatophenyl)propane (Arocy B-10, available from Ciba Geigy AG) and 15 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.16 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 4 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 587 g of methyl ethyl ketone, 273 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) and 172 g of brominated triphenylcyanurate (PYROGUARD SR-245, available from Dai-ich Seiyaku Kogyo Co., Ltd.) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 46 g of p-(α -cumyl)phenol and

of 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a curing accelerator to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0108]

Example 9

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 340 g of toluene and 180 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of α,α' -bis(4-cyanatophenyl)-m-diisopropylbenzene (RTX-366, available from Ciba Geigy AG) and 4 g of p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, iron naphthenate (available from Kanto Kagaku) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 4 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 600 g of methyl ethyl ketone, 330 g of a biphenyl aralkylene novolac type epoxy resin (NC-3000S-H, available from Nippon Kayaku Co., Ltd.) and 164 g of bis(tribromophenoxy)ethane (FF-680, available from Great Lakes) were formulated in the mixture under stirring. Then, after confirmation of dissolution, the mixture was cooled to room temperature, and 23 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a curing accelerator was formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0109]

Example 10

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 360 g of toluene and 225 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of a phenol novolac type cyanate ester resin (REX-371, available from Ciba Geigy AG) and 13 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.15 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 560 g of methyl ethyl ketone, 148 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.), 90 g of a cresol novolac type epoxy resin (ESCN-190-3, available from Sumitomo Chemical Co., Ltd.) and 157 g of hexabromocyclododecane (CD-75P, available from Great Lakes) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 45 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0110]

Example 11

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 290 g of toluene and 225 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and

the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of a phenol novolac type cyanate ester resin (REX-371, available from Ciba Geigy AG) and 13 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.15 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 4 hours to synthesize a phenol-modified cyanate ester oligomer solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 623 g of methyl ethyl ketone, 160 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) and 230 g of a brominated bisphenol A type epoxy resin (ESB400T, available from Sumitomo Chemical Co., Ltd.) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 45 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a curing accelerator were formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0111]

Comparative Example 7

In the same manner as in Example 6 except for removing 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000), and formulating the brominated polystyrene (PDBS-80, available from Great Lakes) with a formulation amount as shown in Table 5, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 6 to prepare a resin varnish having the non-volatile concentration of about 55% by

weight.

[0112]

Comparative Example 8

In the same manner as in Example 6 except for formulating a bisphenol A type epoxy resin (DER-331L, available from Dow Chemical) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with a formulating amount as shown in Table 5, an changing a formulation amount of the brominated polystyrene (PDBS-80, available from Great Lakes) as shown in Table 5, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 6 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0113]

Comparative Example 9

In the same manner as in Example 6 except for formulating p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of p-tert-butylphenol, and a phenol novolac type epoxy resin (N-770, available from Dainippon Ink and Chemicals Incorporated) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether, with formulation amounts as shown in Table 5, respectively, changing the formulation amount of the brominated polystyrene (PDBS-80, available from Great Lakes) to that as shown in Table 5, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 6 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0114]

Comparative Example 10

In the same manner as in Example 7 except for formulating a brominated bisphenol A (TBA, available from Teijin Chemicals, Ltd.) in place of p-tert-octylphenol, a

bisphenol A novolac type epoxy resin (N-865, available from Dainippon Ink and Chemicals Incorporated) in place of a mixed biphenyl type epoxy resin (YL-612H) comprising a biphenol diglycidyl ether and 3,3',5,5'-tetramethylbiphenol diglycidyl ether, and zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) in place of cobalt naphthenate (available from Wako Pure Chemical Industries, Ltd.) with formulation amounts as shown in Table 5, respectively, removing the brominated polyphenylene ether (PO-64P), and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 7 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0115]

Comparative Example 11

In the same manner as in Example 6 except for formulating a brominated bisphenol A (TBA, available from Teijin Chemicals, Ltd.) in place of p-tert-butylphenol, and a phenol salicylic aldehyde novolac type epoxy resin (EPPN-502H, available from Nippon Kayaku Co., Ltd.) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether, with formulation amounts as shown in Table 5, respectively, removing the brominated polystyrene (PDBS-80, available from Great Lakes), and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 6 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0116]

Comparative Example 12

In the same manner as in Comparative Example 10 except for formulating p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of the brominated bisphenol A (TBA), and a dicyclopentadiene skeleton-

containing epoxy resin (HP-7200, available from Dainippon Ink and Chemicals Incorporated) and a polymer type brominated epoxy resin (5203, available from Japan Epoxy Resin Co.) in place of a bisphenol A novolac type epoxy resin (N-685) with formulation amounts as shown in Table 5, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Comparative Example 10 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0117]

[Table 3]

Table 3

Item		Example											Comparative example										
		6	7	8	9	10	11	7	8	9	10	11	7	8	9	10	11	12					
Cyanate ester compound	Aracy B-10	500	-	500	-	-	-	500	500	500	-	-	500	500	500	-	500	-					
	Aracy M-10	-	500	-	-	-	-	-	-	-	-	-	-	-	-	500	-	500					
	RTX-366	-	-	-	450	-	-	-	-	-	-	-	-	-	-	-	-	-					
	REX-371	-	-	-	-	450	450	-	-	-	-	-	-	-	-	-	-	-					
Phenol compound	p-tert-butylphenol	32	-	-	-	-	-	32	32	-	-	-	32	32	-	-	-	-					
	p-tert-octyl phenol	-	47	-	23	45	45	-	-	-	-	-	-	-	-	-	-	-					
	p-(α -cumyl) phenol	-	-	15 46	-	13	13	-	-	-	-	-	-	-	-	-	-	48					
	p-tert-amylphenol	-	-	-	4	-	-	-	-	-	-	-	-	-	12	-	-	-					
Polyphenylene ether resin Epoxy resin	YX-4000	175	100	150	180	225	225	175	175	175	100	175	175	175	175	100	175	100					
	YL-6121H	308	-	273	-	148	160	-	-	-	-	-	-	-	-	-	-	-					
	NC-3000S-H	-	280	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-					
	ESCN-190-3	-	-	-	-	90	-	-	-	-	-	-	-	-	-	-	-	-					
	DER-331L	-	-	-	-	-	-	-	-	-	-	-	-	300	-	-	-	-					
	N-770	-	-	-	-	-	-	-	-	-	-	-	-	-	306	-	-	-					
	N-865	-	-	-	-	-	-	-	-	-	-	-	-	-	-	338	-	-					
	EPPN-502H	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-					
	HP-7200	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	313	-					
	ESB-400T	-	-	-	-	-	230	-	-	-	-	-	-	-	-	-	-	460					

Table 3 (contd.)

Item	Example						Comparative example					
	6	7	8	9	10	11	7	8	9	10	11	12
Flame retardant	207	-	-	-	-	-	147	206	203	-	-	-
	PO-64P	178	-	-	-	-	-	-	-	-	-	-
	SR-245	-	172	-	-	-	-	-	-	-	-	-
	FF-680	-	-	165	-	-	-	-	-	-	-	-
	CD-75P	-	-	-	155	-	-	-	-	-	-	-
	TBA	-	-	-	-	-	-	-	-	205	215	-
Reaction catalyst (Curing accelerator)	EP5203	-	-	-	-	-	-	-	-	-	-	230
	Manganese naphthenate	-	-	0.16	-	0.15	0.15	-	-	-	-	-
	Cobalt naphthenate	-	0.25	-	-	-	-	-	-	-	-	-
	Iron naphthenate	-	-	-	0.14	-	-	-	-	-	-	-
Non-volatile concentration of varnish (% by weight)	Zinc naphthenate	0.13 0.1	0.1	0.1	0.1	0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1
		55	55	55	55	55	55	55	55	55	55	55

[0118]

Resin varnishes obtained in Examples 6 to 11 and

Comparative examples 7 to 12 were each impregnated into a glass cloth (E glass) with a thickness of 0.15 mm, and then, the material was dried at 160°C for 4 to 7 minutes to obtain the respective prepregs with a resin solid content of 52 % by weight. Then, four sheets of this prepregs were laminated, and copper foils with a thickness of each 18 μ m were provided on the both outermost layers, and molding was carried out under the pressing conditions of 230°C for 70 minutes and 2.5MPa to prepare a both-surface copper clad laminated board. With regard to the obtained copper clad laminated board, copper foil peeling strength, dielectric characteristics, solder heat resistance, water absorption rate, flame resistance, bending characteristics, thermal expansion coefficient (α) and Tg (glass transition temperature) were evaluated. The evaluation results are shown in Table 4.

[0119]

Evaluation methods of characteristics of the copper clad laminated boards are as mentioned above. Incidentally, flame resistance of copper clad laminated boards was measured according to UL-94 vertical test method.

[0120]

[Table 4]

Table 4

Item	Example						Comparative example					
	6	7	8	9	10	11	7	8	9	10	11	12
Copper foil peeling strength (kN/m)												
ϵ_r	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.4	1.2	1.2	1.1
1GHz (25°C)	3.51	3.46	3.50	3.42	3.57	3.58	3.51	3.61	3.64	3.67	3.75	3.83
1GHz (90°C)	3.53	3.47	3.51	3.44	3.58	3.59	3.52	3.56	3.70	3.73	3.82	3.92
$\tan \delta$	0.0052	0.0046	0.0052	0.0040	0.0054	0.0055	0.0050	0.0062	0.0067	0.0075	0.0073	0.0075
1GHz (25°C)	0.0057	0.0051	0.0057	0.0045	0.0058	0.0059	0.0053	0.0063	0.0085	0.0092	0.0089	0.0093
1GHz (90°C)	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3
Solder heat resistance (250°C)	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	2/3	1/3	3/3
PCT 1h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	2/3	0/3	0/3
PCT 2h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	0/3
PCT 3h	3/3	3/3	3/3	3/3	3/3	3/3	1/3	1/3	3/3	0/3	0/3	0/3
PCT 4h	3/3	3/3	3/3	3/3	3/3	3/3	0/3	1/3	3/3	0/3	0/3	0/3
PCT 5h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	2/3
(Number of no abnormality /Number of tested)	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	0/3
(238°C)	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	0/3
PCT 1h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	0/3
PCT 2h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	0/3
PCT 3h	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	3/3	0/3	0/3	0/3
PCT 4h	1/3	1/3	2/3	3/3	2/3	2/3	0/3	0/3	0/3	0/3	0/3	0/3
PCT 5h	1/3	0/3	1/3	2/3	1/3	0/3	0/3	0/3	0/3	0/3	0/3	0/3
Water absorption rate (% by weight)	0.52	0.5	0.51	0.47	0.53	0.54	0.5	0.54	0.58	0.73	0.68	0.77
TMA	184	183	185	177	190	190	185	175	188	185	183	167
T _g (°C)	52	52	50	52	51	54	51	54	55	50	61	54
α (ppm/°C, <T _g)	292	295	335	279	313	290	280	356	304	320	289	334
α (ppm/°C, >T _g)	18	18	18	17	19	19	18	18	18	17	17	17
Bending modulus (GPa)	10	9	10	9	12	13	10	7	11	11	10	8
200°C	584	579	598	580	602	593	486	532	521	487	459	487
Room temperature: 25°C	(241)	(215)	(254)	(204)	(283)	(234)	(203)	(125)	(220)	(230)	(276)	(178)
Breaking strength (MPa)	3.7	3.7	3.7	3.8	3.7	3.6	3.0	3.5	3.2	3.0	3.0	3.5
(Yield strength)	(3.8)	(3.6)	(3.7)	(3.7)	(3.6)	(3.6)	(3.2)	(3.5)	(2.3)	(2.1)	(2.0)	(3.3)
Room temperature: 25°C	V-0	V-0	V-0	V-0	V-0	V-0	V-0	V-0	V-0	V-0	V-0	V-0
Breaking elongation (%)												
(Yield strength)												
Flame resistance (UL-94)												

[0121]

As can be clearly seen from Table 4, the laminated boards prepared by using the varnishes of Examples 6 to 11 are excellent in dielectric characteristics (in particular, dielectric dissipation factor) at room temperature (25°C) and 1 GHz than those of the laminated boards of Comparative examples 8 to 12. Moreover, dielectric characteristics (in particular, dielectric dissipation factor) thereof at 90°C are also good, and dependency on temperature change is little as compared to those of Comparative examples 8 to 12. Also, the laminated boards of Examples show lower water absorption rates as compared to those of the laminated boards of Comparative examples 7 to 12 and give better results in solder heat resistance (in particular, 288°C) under humid conditions as compared to the laminated boards of Comparative examples 7 to 12. In addition, the laminated boards of Examples 6 to 11 have higher breaking strength and breaking elongation at room temperature (25°C) and yield elongation at high temperature (200°C) than those of the laminated boards of Comparative examples 7 to 12. Also, excellent characteristics of the laminated boards of these Examples 6 to 11 are accomplished with ensurance of good flame resistance (V-0).

[0122]

[Examples 12 to 16, Comparative examples 13 to 18]

Resin varnishes for metal clad laminated boards were produced according to the formulation amounts as shown in Table 5.

[0123]

Example 12

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 300 g of toluene and 175 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of 2,2-bis(4-cyanatophenyl)propane (Arocy B-10, available from Ciba Geigy AG) and 32 g of p-

tert-butylphenol (available from Kanto Kagaku) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.13 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester resin-containing solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 530 g of methyl ethyl ketone and 308 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (a tetramethylbiphenyl type epoxy resin, YX-4000, available from Japan Epoxy Resin Co.) were formulated in the mixture under stirring. Then, after confirmation of dissolution, the mixture was cooled to room temperature, 3.0 g of 2,6-di-tert-butyl-4-methylphenol (TBMP) as an antioxidant, and 0.1 g of zinc naphthenate as a curing accelerator were formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0124]

Example 13

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 285 g of toluene and 150 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 500 g of 2,2-bis(4-cyanatophenyl)propane (Arocy B-10, available from Ciba Geigy AG) and 15 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.16 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester resin-containing solution which is compatibilized with a poly-

phenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 522 g of methyl ethyl ketone and 273 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 46 g of p-(α -cumyl)phenol, 3.0 g of 4,4'-butylidenebis(3-methyl-6-tert-butylphenol) (BBMTBP) as an antioxidant, and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a curing accelerator were formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0125]

Example 14

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 290 g of toluene and 225 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of a phenol novolac type cyanate ester resin (REX-371, available from Ciba Geigy AG) and 13 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.15 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 2 hours to synthesize a phenol-modified cyanate ester resin-containing solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 500 g of methyl ethyl ketone, 148 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) and 90 g of a cresol novolac type epoxy resin (ESCN-190-3, available from Sumitomo Chemical Co.,

Ltd.) were formulated in the mixture under stirring, and after confirmation of dissolution and the mixture was cooled to room temperature, 45 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.), 2.7 g of dilaurylthiodipropionate as an antioxidant, and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a curing accelerator were formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0126]

Example 15

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 270 g of toluene and 180 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of α,α' -bis(4-cyanatophenyl)-m-diisopropylbenzene (RTX-366, available from Ciba Geigy AG) and 4 g of p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, of 0.14 g iron naphthenate (available from Kanto Kagaku) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 3 hours to synthesize a phenol-modified cyanate ester resin-containing solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 540 g of methyl ethyl ketone and 330 g of a biphenyl aralkylene novolac type epoxy resin (NC-3000S-H, available from Nippon Kayaku Co., Ltd.) were formulated in the mixture under stirring. Then, after confirmation of dissolution and the mixture was cooled to room temperature, 23 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.), 3.0 g of 4,4'-butylidenebis(3-methyl-6-tert-butylphenol) (BBMTBP) as an antioxidant, and 0.1 g of zinc naphthenate (available

from Wako Pure Chemical Industries, Ltd.) as a curing accelerator were formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0127]

Example 16

In a 3-liter four-necked separable flask equipped with a thermometer, a condenser and a stirring device, 290 g of toluene and 225 g of polyphenylene ether resin (PKN4752, available from GE Japan Co.) were charged, and the mixture was heated to 90°C and dissolved under stirring. Next, 450 g of a phenol novolac type cyanate ester resin (REX-371, available from Ciba Geigy AG) and 13 g of p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) were charged, after confirmation of dissolution and the liquid temperature was maintained at 110°C, 0.15 g of manganese naphthenate (available from Wako Pure Chemical Industries, Ltd.) was formulated in the mixture as a reaction promoter, and the mixture was reacted for about 2 hours to synthesize a phenol-modified cyanate ester resin-containing solution which is compatibilized with a polyphenylene ether resin. Then, the reaction solution was cooled, and when the inner temperature reached to 80°C, 500 g of methyl ethyl ketone, 148 g of 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000, available from Japan Epoxy Resin Co.) and 90 g of a cresol novolac type epoxy resin (ESCN-190-3, available from Sumitomo Chemical Co., Ltd.) were formulated in the mixture under stirring, and after confirmation of dissolution and the mixture was cooled to room temperature, 45 g of p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.), 2.7 g of dilaurylthiodipropionate as an antioxidant, and 0.1 g of zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) as a curing accelerator were formulated in the mixture to prepare a resin varnish having a non-volatile concentration of about 55% by weight.

[0128]

Comparative Example 13

In the same manner as in Example 12 except for removing 3,3',5,5'-tetramethylbiphenol diglycidyl ether (YX-4000) and 2,6-di-tert-butyl-4-methylphenol (TBMP), and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 12 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0129]

Comparative Example 14

In the same manner as in Example 12 except for removing 2,6-di-tert-butyl-4-methylphenol (TBMP), and formulating p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of p-tert-butylphenol, a phenol novolac type epoxy resin (N-770, available from Dainippon Ink and Chemicals Incorporated) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with formulation amounts as shown in Table 7, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 12 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0130]

Comparative Example 15

In the same manner as in Example 12 except for removing 2,6-di-tert-butyl-4-methylphenol (TBMP), and formulating p-tert-amylphenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of p-tert-butylphenol, a phenol novolac type epoxy resin (N-770, available from Dainippon Ink and Chemicals Incorporated) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with formulation amounts as shown in Table 7, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a

non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 12 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0131]

Comparative Example 16

In the same manner as in Example 13 except for removing 2,2'-methylene-bis-(4-methyl-6-tert-butylphenol) (MBMTBP), formulating a bisphenol A novolac type epoxy resin (N-865, available from Dainippon Ink and Chemicals Incorporated) in place of the mixed biphenyl type epoxy resin (YL-612H) comprising biphenol diglycidyl ether and 3,3',5,5'-tetramethylbiphenol diglycidyl ether, and zinc naphthenate (available from Wako Pure Chemical Industries, Ltd.) in place of cobalt naphthenate (available from Wako Pure Chemical Industries, Ltd.) with formulation amounts as shown in Table 5, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 13 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0132]

Comparative Example 17

In the same manner as in Example 12 except for removing 2,6-di-tert-butyl-4-methylphenol (TBMP), formulating p-tert-octylphenol (available from Wako Pure Chemical Industries, Ltd.) in place of p-tert-butylphenol, and a phenol salicylic aldehyde novolac type epoxy resin (EPPN-502H, available from Nippon Kayaku Co., Ltd.) in place of 3,3',5,5'-tetramethylbiphenol diglycidyl ether with formulation amounts as shown in Table 5, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Example 12 to prepare a resin varnish having the non-volatile concentration of

about 55% by weight.

[0133]

Comparative Example 18

In the same manner as in Comparative example 16 except for formulating p-(α -cumyl)phenol (available from Tokyo Kasei Kogyo Co., Ltd.) in place of p-tert-octylphenol and a dicyclopentadiene skeleton-containing epoxy resin (HP-7200, available from Dainippon Ink and Chemicals Incorporated) in place of a bisphenol A novolac type epoxy resin (N-685) with formulation amounts as shown in Table 5, respectively, and methyl ethyl ketone was so formulated in a formulation amount that a non-volatile concentration became about 55% by weight, the procedure was carried out as in Comparative example 16 to prepare a resin varnish having the non-volatile concentration of about 55% by weight.

[0134]

[Table 5]

Table 5

Item		Example							Comparative example						
		12	13	14	15	16	13	14	15	16	17	18			
Cyanate ester compound	Atocry B-10	500	-	500	-	-	500	500	500	500	-	500	-		
	Atocry M-10	-	500	-	-	-	-	-	-	-	500	-	500		
	RTX-366	-	-	-	450	-	-	-	-	-	-	-	-		
	REX-371	-	-	-	-	450	-	-	-	-	-	-	-		
Phenol compound	p-tert-butylphenol	32	-	-	-	-	-	32	32	-	-	-	-		
	p-tert-octylphenol	-	47	-	23	45	-	-	-	-	47	44	-		
	p-(α-cumyl) phenol	-	-	15 46	-	13	-	-	-	-	-	-	48		
	p-tert-amylphenol	-	-	-	4	-	-	-	-	12	-	-	-		
Polyphenylene ether resin	175	100	150	180	225	175	175	175	175	100	175	100			
Epoxy resin	YX-4000	308	-	273	-	148	-	-	-	-	-	-	-		
	YL-6121H	-	280	-	-	-	-	-	-	-	-	-	-		
	NC-3000S-H	-	-	-	330	-	-	-	-	-	-	-	-		
	ESCH-190-3	-	-	-	-	90	-	-	-	-	-	-	-		
	DER-331L	-	-	-	-	-	-	300	-	-	-	-	-		
	N-770	-	-	-	-	-	-	-	306	-	-	-	-		
	N-865	-	-	-	-	-	-	-	-	338	-	-	-		
	EPPN-502H	-	-	-	-	-	-	-	-	-	313	-	-		
	HP-7200	-	-	-	-	-	-	-	-	-	-	460	-		
	TBMP	3.0	-	-	-	-	-	-	-	-	-	-	-		
	MBMTBP	-	2.8	-	-	-	-	-	-	-	-	-	-		
	BBMTBP	-	-	3.0	3.0	-	-	-	-	-	-	-	-		
Reaction catalyst (Curing accelerator)	Dilaurylthiodipropionate	-	-	-	-	2.7	-	-	-	-	-	-	-		
	Manganese naphthenate	-	-	0.16	-	0.15	-	-	-	-	-	-	-		
	Cobalt naphthenate	-	0.25	-	-	-	-	-	-	-	-	-	-		
	Iron naphthenate	-	-	-	0.14	-	-	-	-	-	-	-	-		
Non-volatile concentration of varnish (% by weight)	Zinc naphthenate	0.13 0.1	0.1	0.1	0.1	0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1	0.13 0.1		
		55	55	55	55	55	55	55	55	55	55	55	55		

[0135]

Resin varnishes obtained in Examples 12 to 16 and Comparative examples 13 to 18 were each impregnated into a glass cloth (E glass) with a thickness of 0.15 mm, and then, the material was dried at 160°C for 4 to 7 minutes to obtain the respective prepregs with a resin solid content of 52 % by weight. Then, four sheets of this prepregs were laminated, and copper foils with a thickness of each 18 μm were provided on the both outermost layers, and molding was carried out under the pressing conditions of 230°C for 70 minutes and 2.5MPa to prepare a both-surface copper clad laminated board. With regard to the obtained copper clad laminated board, copper foil peeling strength, dielectric characteristics, solder heat resistance, water absorption rate, flame resistance, bending characteristics, thermal expansion coefficient (α) and Tg (glass transition temperature) were evaluated. The evaluation results are shown in Table 6.

[0136]

Evaluation methods of characteristics of the copper clad laminated boards are as mentioned above.

Electrical corrosion resistance of the copper clad laminated boards was carried out by preparing a test pattern wiring board which had been prepared by penetrating through holes with a hole-distance of 350 μm (drilling conditions; rotation number: 80,000 rpm, feed rate: 2,400 mm/min) to a copper clad laminated board using a drill having a diameter of 0.4 mm, and then, subjecting to through hole plating according to the conventional manner. To the respective test pieces was applied 100 V under 85°C/85% RH atmosphere and a time until connection breakage occurs was measured.

[0137]

[Table 6]

Table 6

Item	Example						Comparative example					
	12	13	14	15	16		13	14	15	16	17	18
Copper foil peeling strength (kN/m) $\epsilon \tau$	1.5	1.5	1.5	1.5	1.5		1.5	1.5	1.4	1.4	1.4	1.4
	3.50	3.45	3.50	3.41	3.57		3.50	3.50	3.63	3.52	3.60	3.50
	3.52	3.46	3.52	3.42	3.58		3.51	3.65	3.69	3.59	3.66	3.58
	0.0053	0.004	0.0052	0.0040	0.0054		0.0050	0.0062	0.0067	0.0065	0.0066	0.0054
$\tan \delta$	0.0057	0.0051	0.0057	0.0045	0.0058		0.0053	0.0083	0.0085	0.0082	0.0080	0.0066
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	2/3	3/3	3/3	3/3	3/3
Solder heat resistance (260°C) (Number of no abnormality /Number of tested) (288°C)	3/3	3/3	3/3	3/3	3/3		3/3	1/3	3/3	3/3	3/3	1/3
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	1/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	1/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
	3/3	3/3	3/3	3/3	3/3		3/3	3/3	3/3	3/3	3/3	3/3
Water absorption rate (% by weight)	1/3	2/3	2/3	3/3	2/3		0/3	0/3	0/3	0/3	0/3	0/3
	1/3	0/3	0/3	3/3	1/3		0/3	0/3	0/3	0/3	0/3	0/3
	0.52	0.50	0.52	0.48	0.54		0.50	0.54	0.58	0.58	0.62	0.53
	185	186	184	177	195		185	175	188	191	191	185
TMA	51	54	48	55	51		50	55	51	51	63	59
	301	303	294	289	302		285	377	322	328	278	350
Bending modulus(GPa)	18	18	18	17	19		18	18	18	18	18	18
	10	9	10	9	13		11	7	12	13	13	9
Breaking strength(MPa)	586	579	600	578	626		488	558	516	502	512	536
	(239)	(231)	(229)	(204)	(287)		(210)	(142)	(244)	(232)	(286)	(189)
Breaking elongation(%)	3.7	3.7	3.9	3.7	3.8		3.0	3.5	3.2	3.2	3.2	3.4
	(3.7)	(3.8)	(3.8)	(3.8)	(3.7)		(3.5)	(3.5)	(2.4)	(2.3)	(2.2)	(3.5)
Electrical corrosion resistance (Number of days until connection breakage)	>500	>500	>500	>500	>500		240	288	168	72	36	144

[0138]

As can be clearly seen from Table 6, the laminated boards prepared by using the varnishes of Examples 12 to 16 are excellent in dielectric characteristics (in particular, dielectric dissipation factor) at room temperature (25°C) and 1 GHz than those of the laminated boards of Comparative examples 14 to 17. Moreover, dielectric characteristics (in particular, dielectric dissipation factor) thereof at 90°C are also good, and dependency on temperature change is little as compared to those of Comparative examples 14 to 18. Also, the laminated boards of Examples show lower water absorption rates as compared to those of the laminated boards of Comparative examples 14 to 18 and give better results in solder heat resistance (in particular, 288°C) under humid conditions as compared to the laminated boards of Comparative examples 13 to 18. In addition, the laminated boards of Examples 12 to 16 have higher breaking strength and breaking elongation at room temperature (25°C) and yield elongation at high temperature (200°C) than those of the laminated boards of Comparative examples 13 to 18. Also, the laminated boards of Examples 12 to 16 show better electrical corrosion resistance as compared with those of the laminated boards of Comparative example.

[0139]

[Effects of the invention]

Accordingly, it could be found that cured products using the resin composition for printed wiring board of the present invention are excellent in dielectric characteristics at high frequency band region, and a drifting property of dielectric characteristics due to change in temperature is also little. Also, they have high bending strength and elongation at glass state region and elongation at high temperatures region. Moreover, the metal clad laminated boards prepared by using the present resin composition for printed wiring board containing a flame retardant gives good flame resistance in addition to these excellent characteristics. Also, the present resin

composition for printed wiring board containing an antioxidant gives good electric corrosion resistance in addition to these excellent characteristics. Accordingly, they can be expected to be members or parts for a printed wiring board to be used for various kinds of electric and electronic devices which treats high frequency signals of 1 GHz or higher. In particular, the excellent high frequency characteristics and bending characteristics, and high humidity and heat resistance which are characteristics of the resin composition of the present invention are effective for the uses of laminated boards and prepregs to be used for a multi-layered printed wiring board such as a high speed server, router and high speed base station devices, which are required to have heat resistance or crack resistance under severe conditions.

[Document name] Abstract

[Summary]

[Problem] The present invention provides a resin composition for printed wiring board to be used for electronic devices in which operating frequency exceeds 1 GHz, and a varnish, a prepreg and a metal clad laminated board using the same.

[Solving means] A resin composition for printed wiring board characterized in that it is obtained by using (A) a cyanate ester compound having 2 or more cyanate groups in the molecule and/or a prepolymer thereof, (B) a monovalent phenol compound, (C) a polyphenylene ether resin, and (D) an epoxy resin containing at least one kind of an epoxy resin having a biphenyl structure in the molecule, and a varnish, a prepreg and a metal clad laminated board using the same.

[Selective figure] None